

Behavior of Organic Sources of Phosphorus in Agricultural Systems

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Plant Available Phosphorus

⊗ Inorganic orthophosphates



Organic Phosphorus

- ❖ Surface Soils (75% of Total P)
- ❖ Average C:N:P ratio 140:10:1.3)
 - ❖ C:P range
 - ❖ 67 -311:1 Virgin Soils
 - ❖ 141-526:1 Cultivated Soils

Factors Influencing C:P Ratio

- P Supply
- Parent Materials
- Texture
- Climate
- pH
- Depth of Soil

Mineralization of Organic P

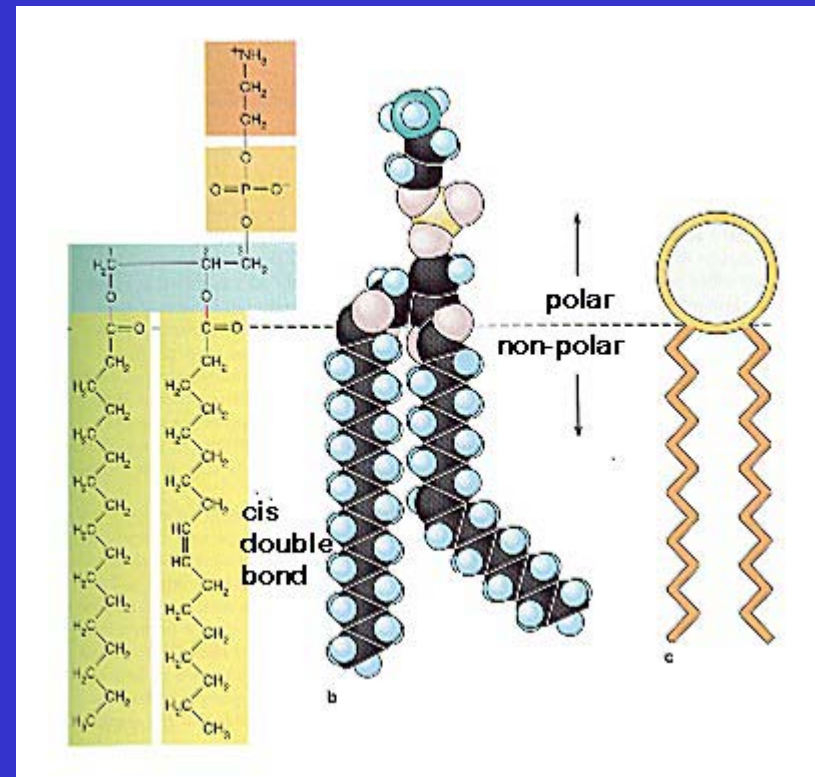
- Organic P \longrightarrow Inorganic P
- C:P < 200 Mineralization (P gain)
- C:P 200 - 300 No Change
- C:P > 300 Immobilization (P loss)

Factors Controlling Mineralization of P

- Chemical form of organic P
- pH
- Temperature
- Soil Moisture
- Soil Microbial Communities
- Soil Enzymes

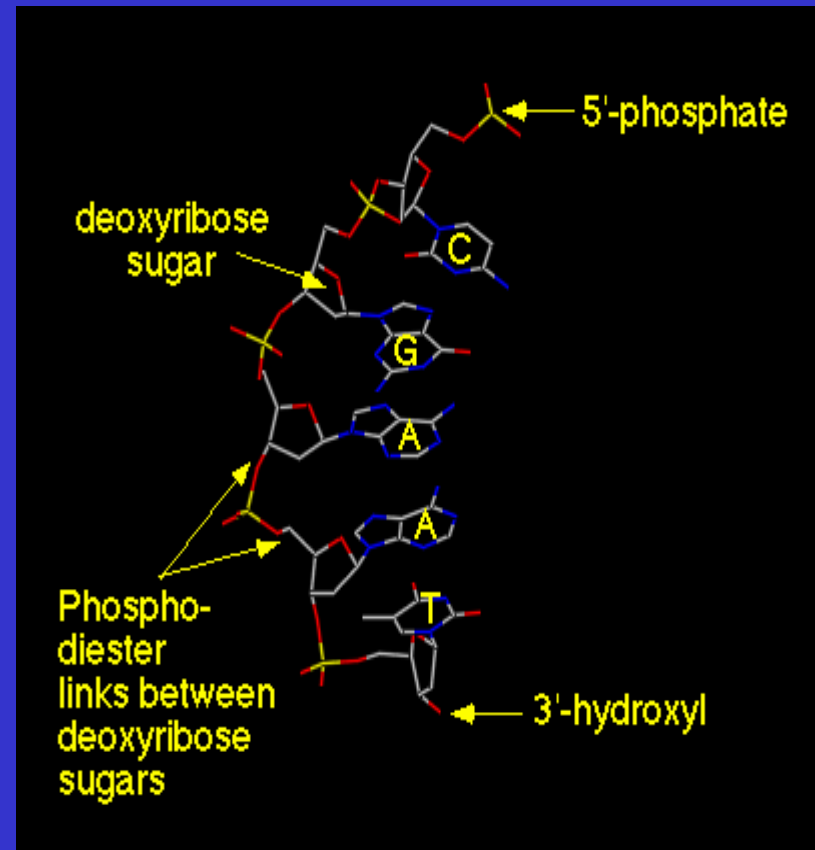
Phospholipids

- Parts of Cell Membranes
- Ill defined Group
- Estimates 1-3% of Soil Organic Matter
- Underestimated
- Strongly associated with soil mineral fractions



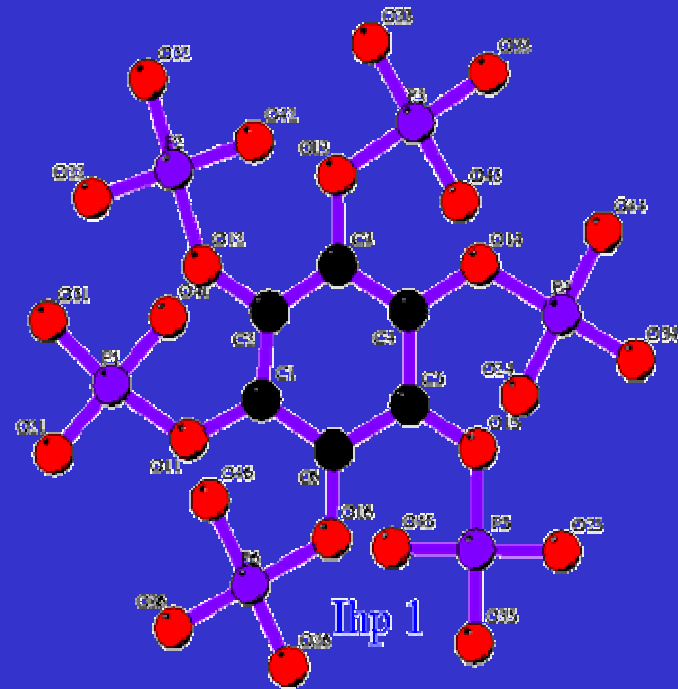
Nucleic Acids

- Present in all Cells
- Backbone of Nucleic acids
- Easily Decomposed
- Less than 3% of Soil Organic Phosphorus

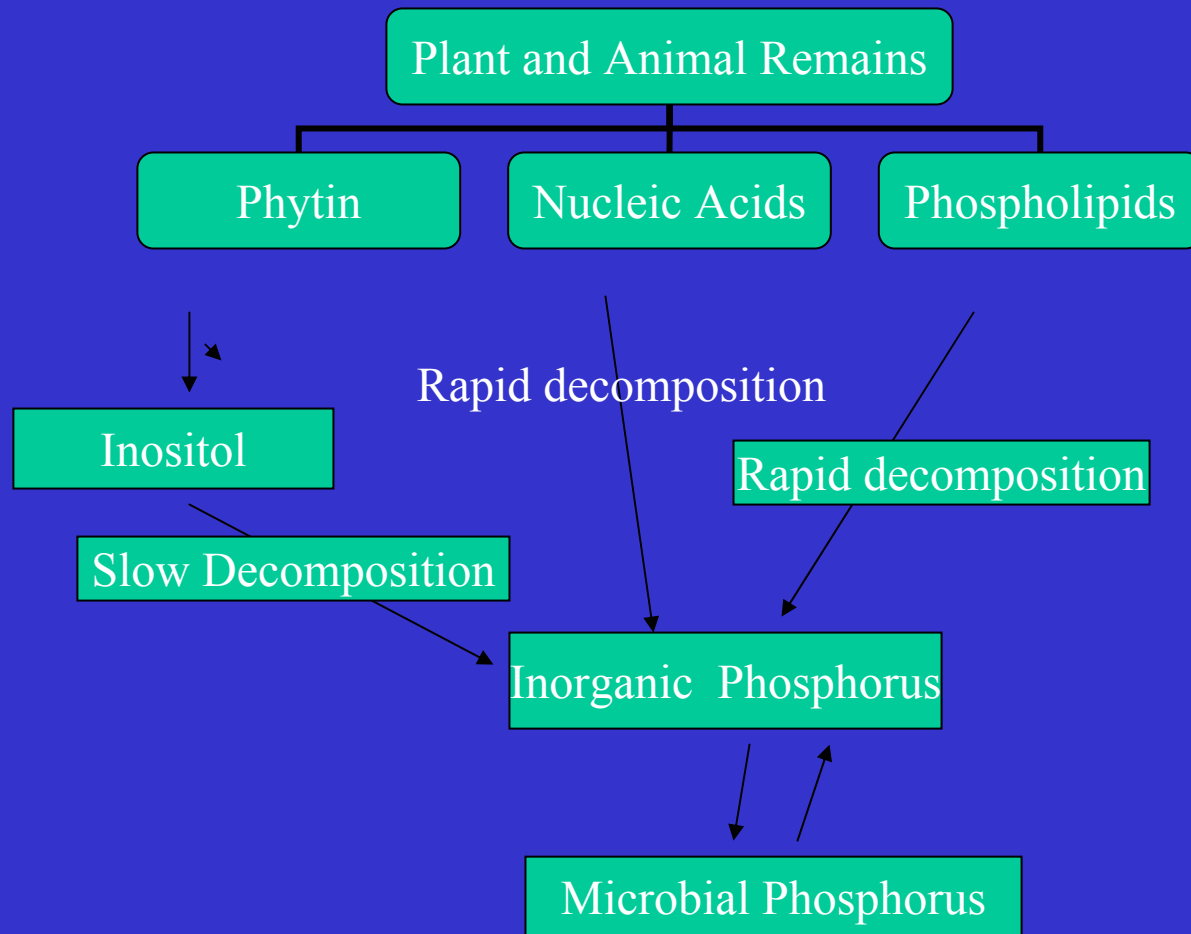


Phytic Acid (Inositol Hexaphosphate)

- Mixed with Ca
- Cereal Grains
- Insoluble complexes
- Slow Decomposition
- High Stability
- Limited Phytase in Soils
- Synthesized in Soils



Mineralization of Organic Phosphorus



Distribution of Inositol P in Soils

Form	% of Soil Organic P	Percentage of IP
Mono	1 -6	1-11
Di and Tri	4-11	6-15
Tetra	5-18	8-30
Penta	15-27	15-30
Hexa	20-66	20-66

Recovery of Organic P

Soil	Organic P ppm	IHP %	Phospholipids %	Nucleic Acids%	Total %
1	39	44	7	2	54
2	187	66	2	0.2	68
3	250	60	1	0.5	61

Water Soluble Organic P

P form	P Fraction mg/kg	%
Total Organic P	779	100
Monoester P	55	7
Inositol P	302	40
Nucleic Acids	43	6
Unreleased	377	48

NaHCO₃ Extractable Organic P in Manure

P form	P Faction	%
Total Organic P	426	100
Monoester P	1	3
Inositol P	14	3
Nucleic Acids	5	1.1
Unreleased	48	86

Mode Of Organic P Loss

- ❖ Two forms: soluble P and particulate P.
 - Soluble P:
 - Inositol hexaphosphate
 - Subsurface flow in tile drainage systems;
 - Not directly available to algae.

Modes Of P Loss (Contd.)

- ❖ Particulate P (Sediment P):
 - rides fine-sized particles;
 - less available;
 - forms better part of total P loss.

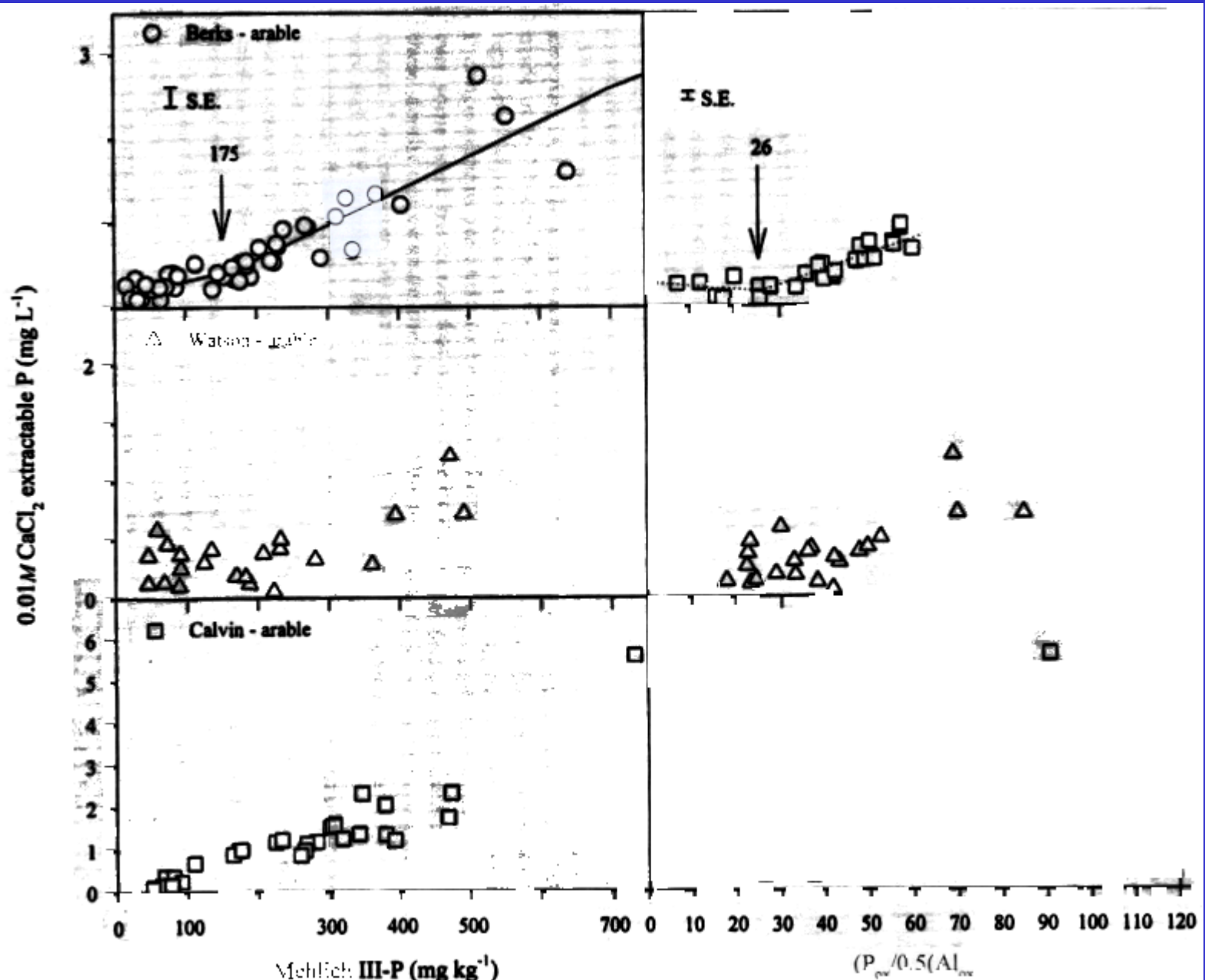
Lack Of Control Of Dissolved P

- ◆ Most of the management practices, namely, riparian buffer strips, conservation tillage, terracing, contour plowing, impoundments, etc., effective in controlling particulate P, but do not reduce soluble P levels adequately.

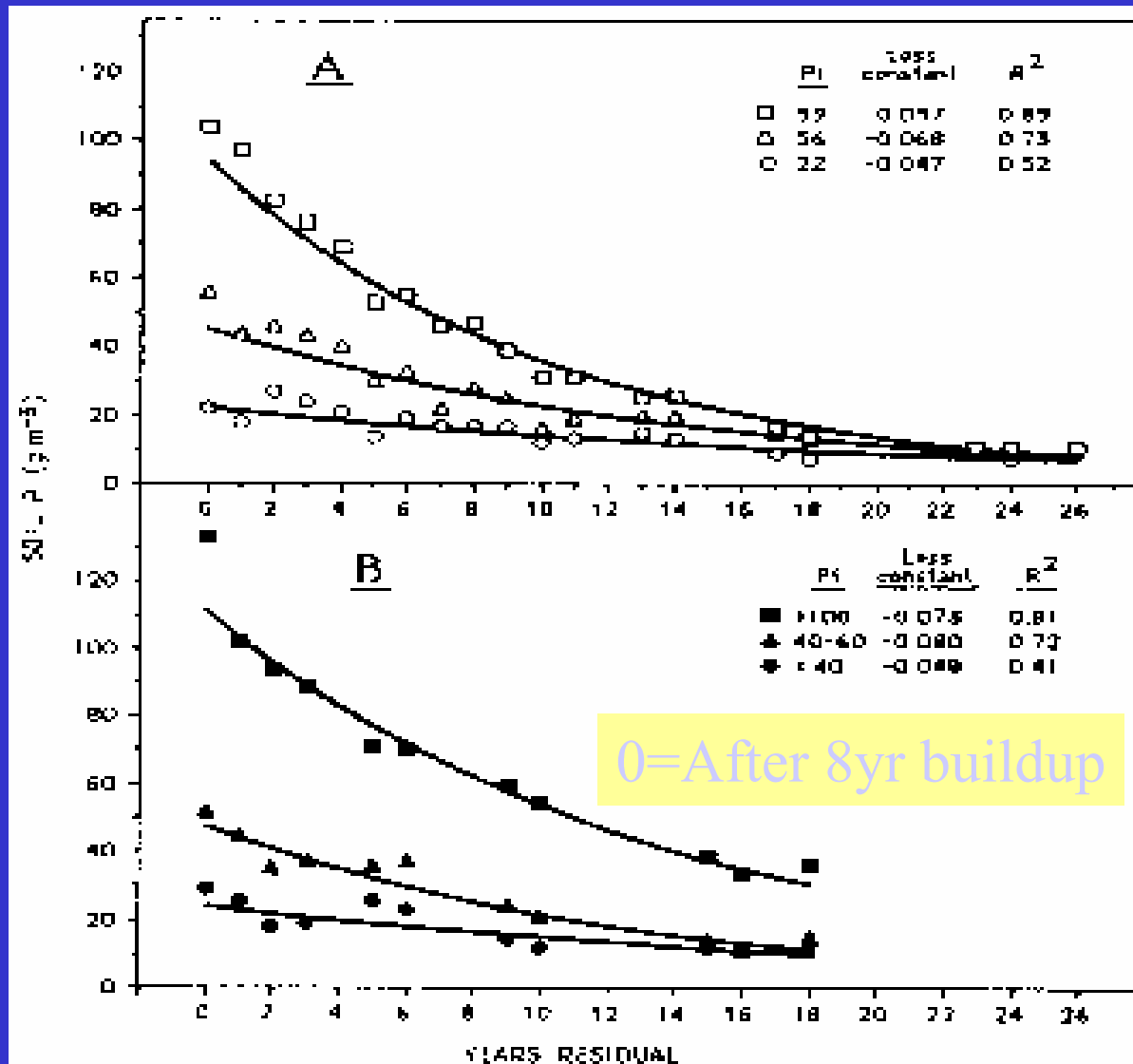
Loss Of Dissolved P:relation Between Soil Test P And Water Soluble P

- ★ With increase in soil test P (STP), water soluble P(or the propensity of the soil to release P) in the runoff increases.
- ★ Release explained by two linear relations of significantly different slopes on either side of a point, called change-point;

Change-point



Slow crop removal (McCollum, 1991)



**Crop removal of residual P
doesn't have
adequate effect**

THEREFORE, NEED TO CONTROL DISSOLVED

P

How ??

- ✧ Reducing dissolved P content in manures.
- ✧ Increasing P sorption capacity(or decreasing degree of phosphorus saturation) of the soil or border strips.

Iron And Aluminum Oxides

- ❑ Have great affinity for dissolved P.
- ❑ Several industrial and mining wastes are rich in these oxides.
- ❑ Acid mine drainage floc is one.

Use Of Industrial And Mining Wastes

- ❖ Peacock and Rimmer (2000) examined the suitability of an iron oxide-rich gypsum byproduct as a soil amendment.
- ❖ Red gypsum: a byproduct of titanium oxide pigment manufacturing industry; contains about 35% iron oxides.
- ❖ P sorption characteristics of red gypsum compared with those of pure gypsum and iron oxides isolated from the red gypsum.

**So, other readily available
iron oxide-rich mining wastes
should be evaluated
for their
ability to attenuate P
in runoff and leachates
from soils**



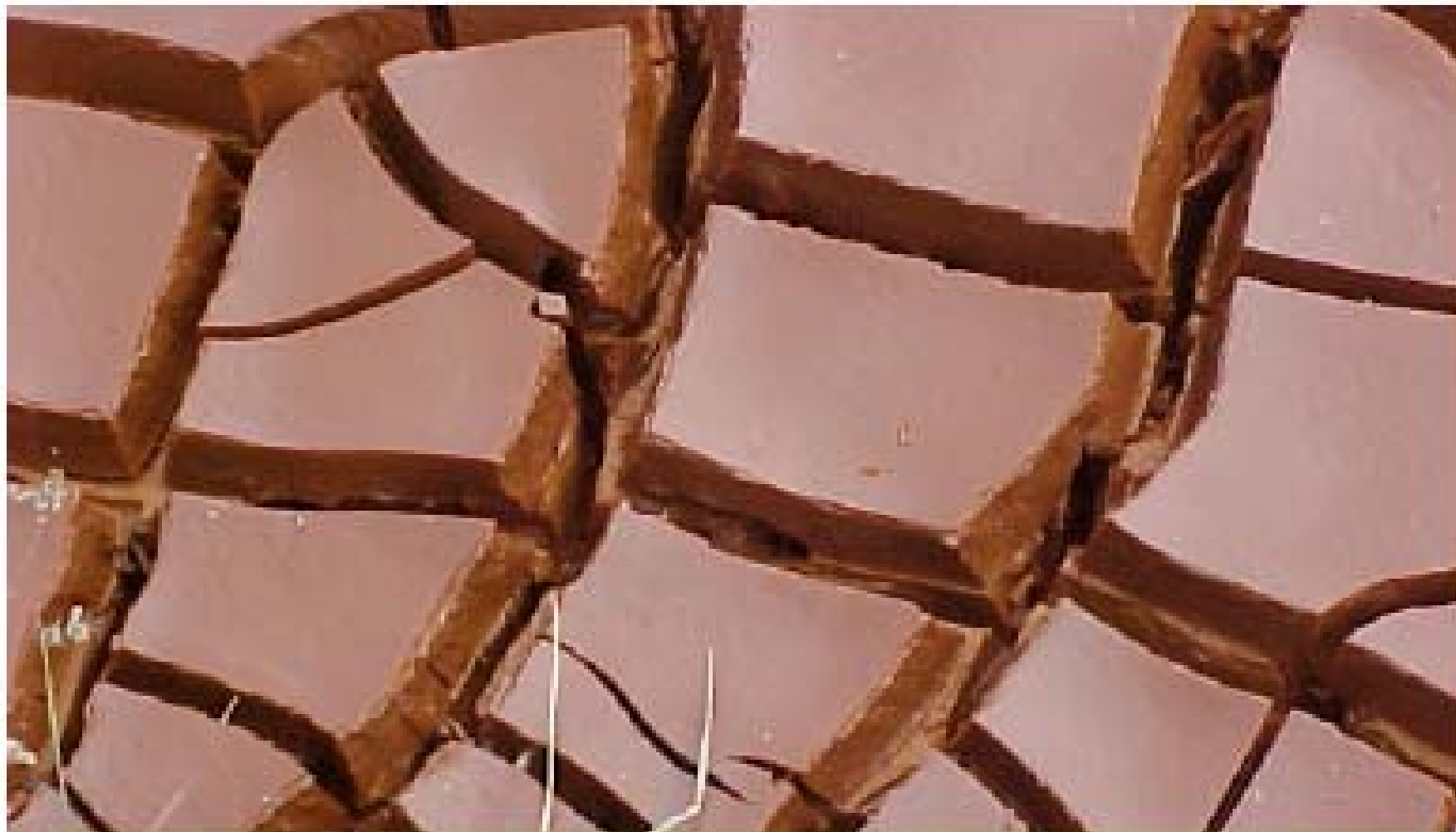


Fig. 2. Initial drying stage of acid mine drainage floc (5% solids)

Objectives

- Study of effectiveness of border strips treated with acid mine drainage (AMD) floc in controlling runoff P loss; and
- ❖ Study of P sorption onto AMD floc.

Sorption of P onto acid mine drainage floc

- Objectives:
 - To assess P remediation potential of various AMD flocs.
 - To investigate implications of P sorption on various AMD flocs in determining environmental thresholds of P for soils in which amorphous Fe and Al oxides are the main controllers of P movement.

Materials and Methods

- AMD collected from Omega mine in Monongalia county of WV.
- One part of AMD collected and titrated at the site of discharge with 1.0M ammonium hydroxide to pH 7.
- Similarly, the other part titrated with 1.0M NaOH.
- The third type of floc was lime-treated.

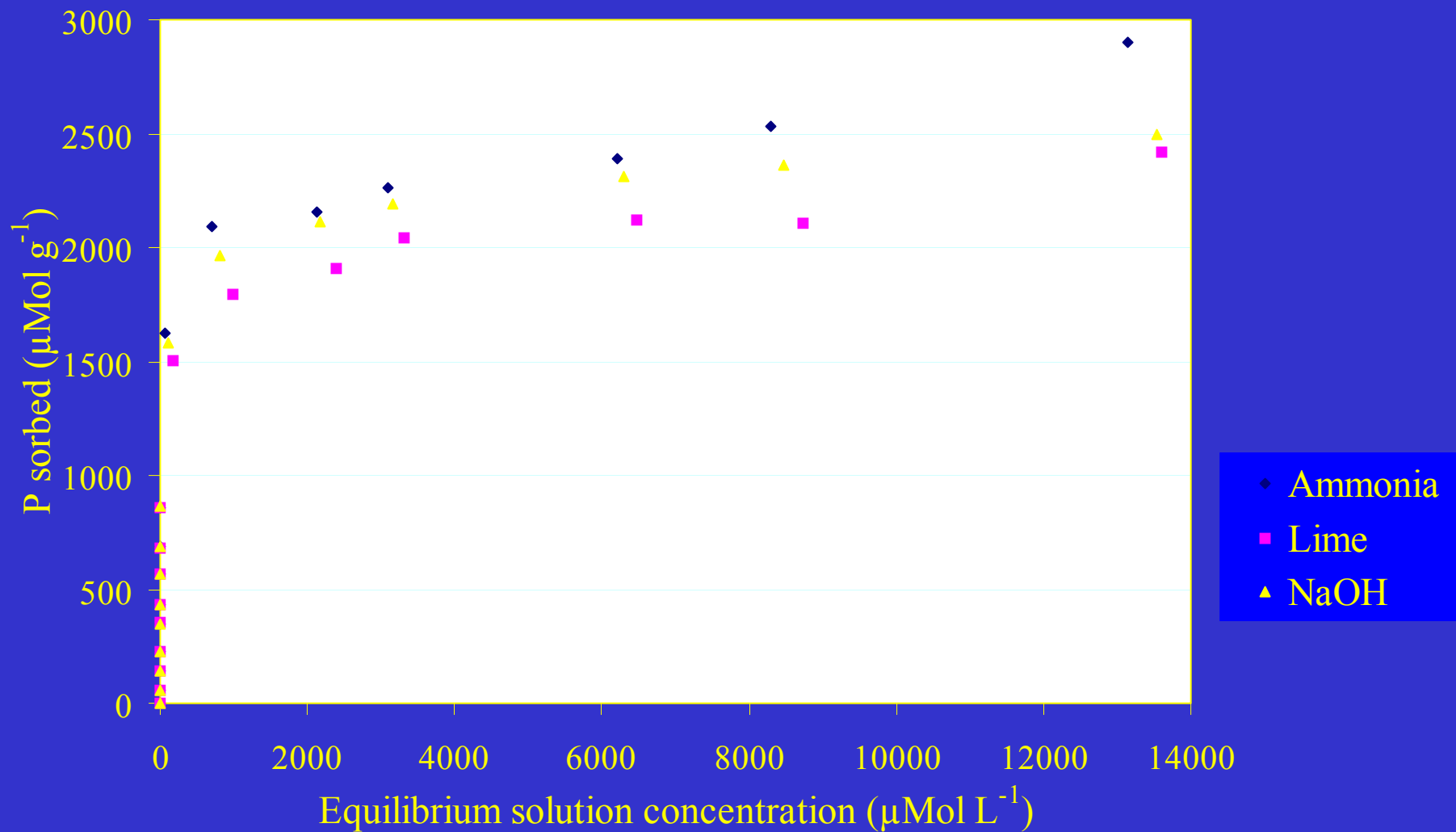
Materials and Methods (contd.)

- Similarly, another set equilibrated with inositol hexaphosphate (IHP)-P.
- Three replicates.
- Measurements for inorganic P, IHP-P and metals on ICP-AES.

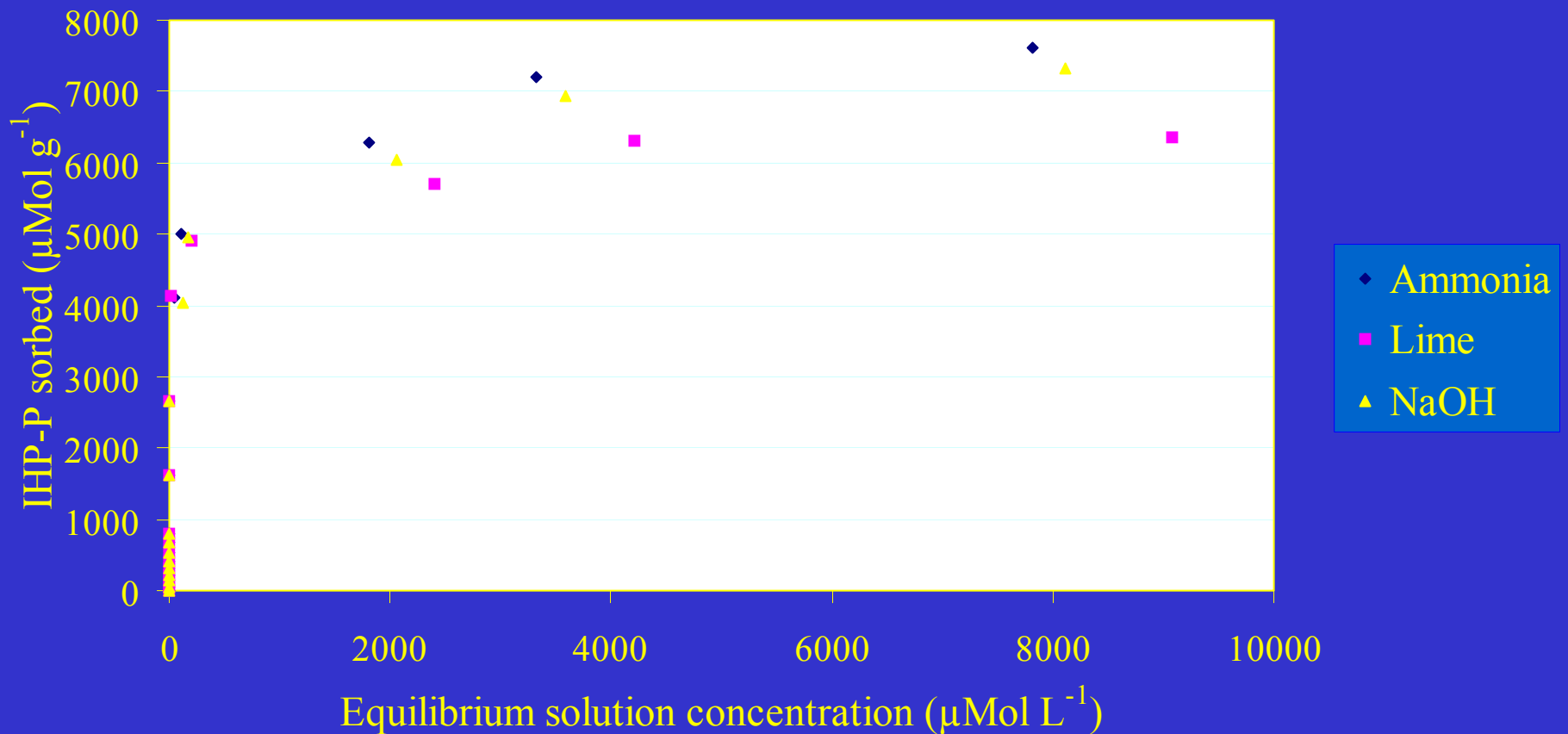
Elemental composition and P sorption capacity of the flocs

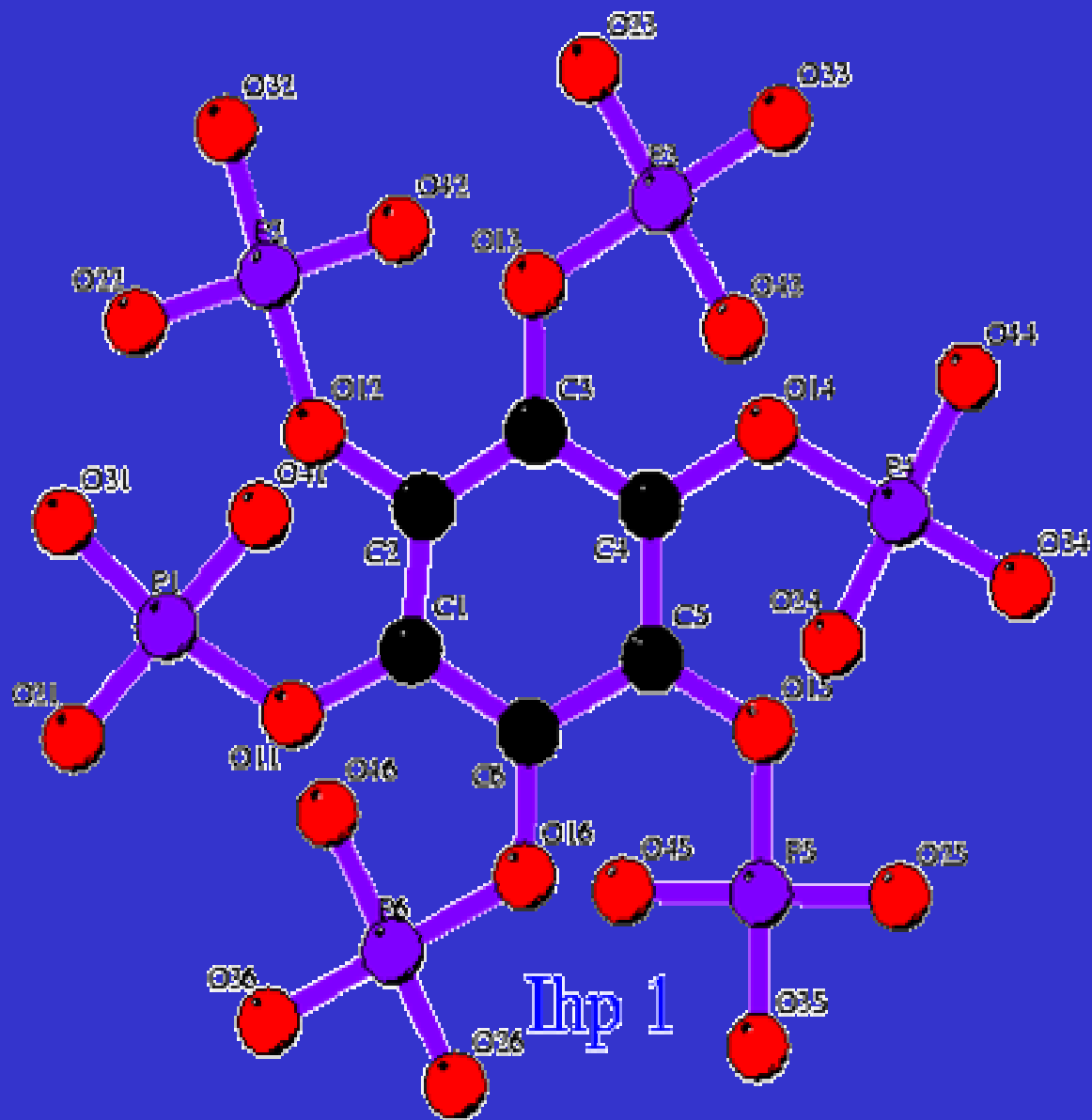
- >99% of total Fe amorphous.
- > 98% of total Al amorphous.

Sorption of inorganic-P on various acid mine drainage floccs

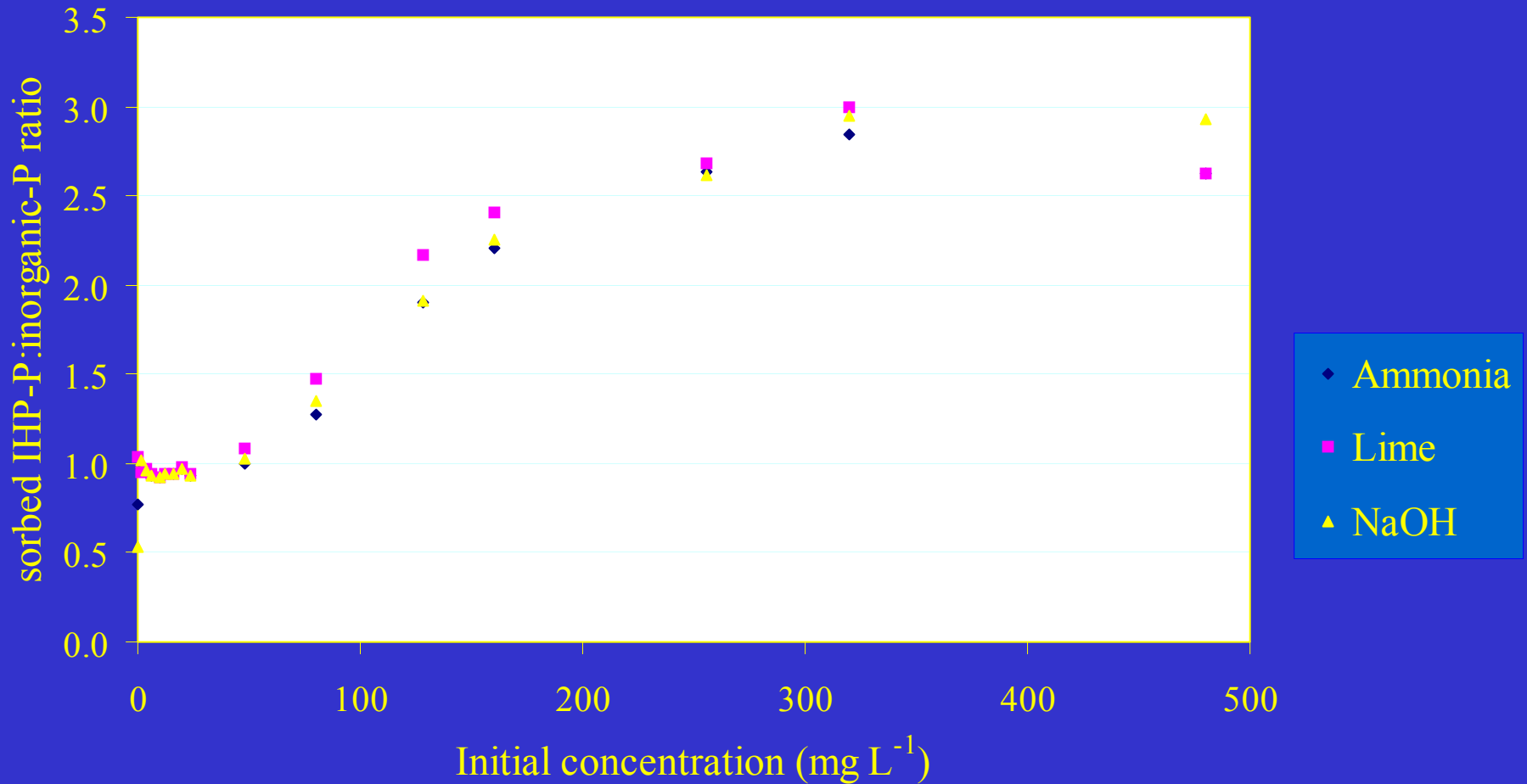


Sorption of inositol hexaphosphate (IHP)-P on various acid mine drainage flocs.





Ratio of inositol hexaphosphate (IHP)-P and inorganic P sorbed onto various acid mine drainage flocs.



Conclusions

- Various AMD floccs can sorb substantial amounts of P.
- Far greater amounts of IHP-P than inorganic-P sorbed.

Hydrolysis of inositol hexaphosphate (IHP) by AMD floc

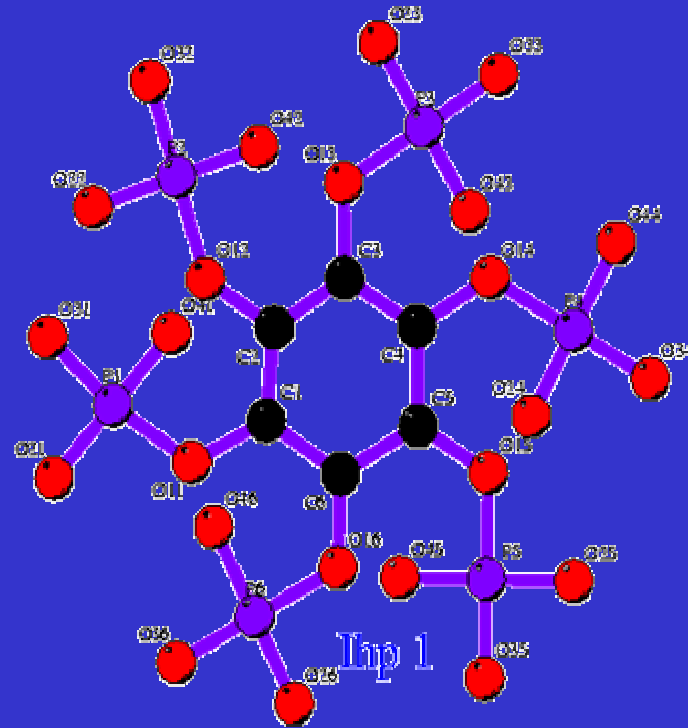
- Inositol hexaphosphate (IHP) or phytate sorbs strongly onto Fe and Al oxides.
- Some studies: metal oxides facilitate hydrolysis of organic compounds.

Consequences of hydrolysis

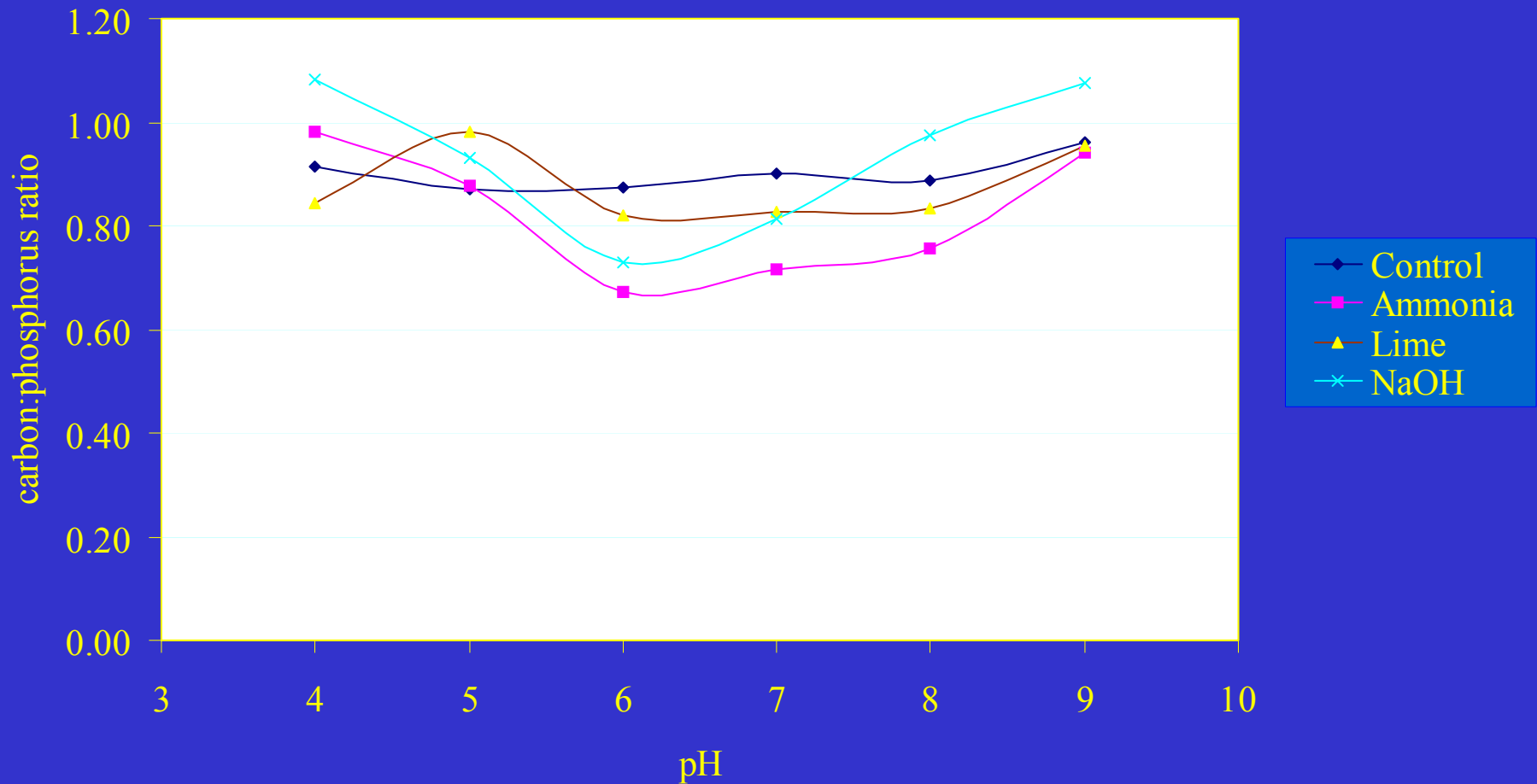
- Hypothesis: P retained, organic moiety released.
- Then, P attenuation capacity of flocs impaired.
- If IHP hydrolyzed first, P ions take more sites.

Diagnosis for hydrolysis

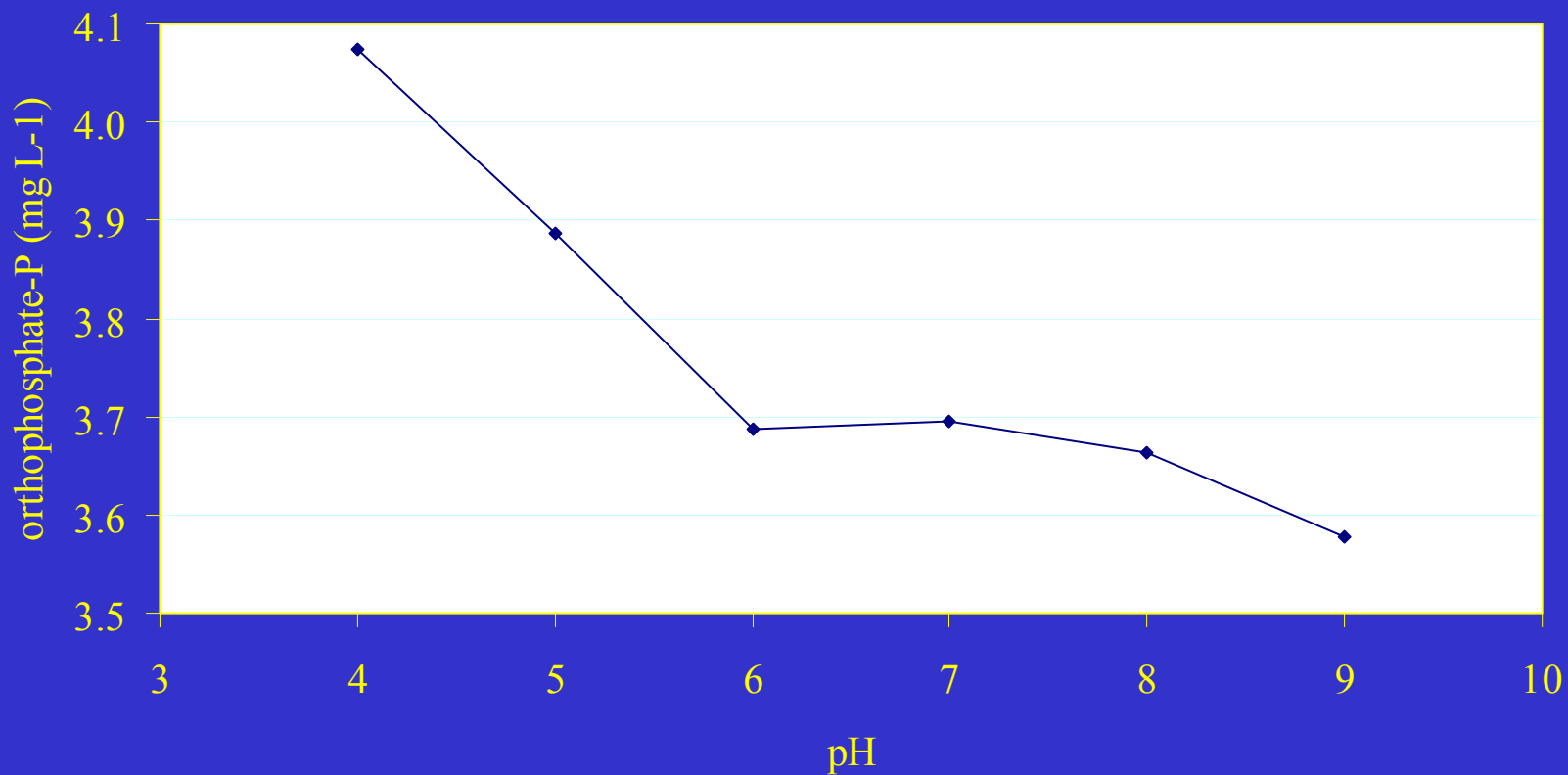
- Carbon: Phosphorus ratio in equilibrium solution (with AMD floc).
- If ratio more than that in blank, IHP-P sorbed after hydrolysis.
- If ratio same: sorbed as it is.



Carbon: Phosphorus ratio of IHP-P solutions equilibrated with or without AMD floccs.



Orthophosphate-P concentrations of 200mg L⁻¹ IHP-P solution at various pH levels



Conclusions

- IHP sorbed *as such*.
- No appreciable effect of pH.
- AMD floc attenuates much more IHP-P than inorganic P.

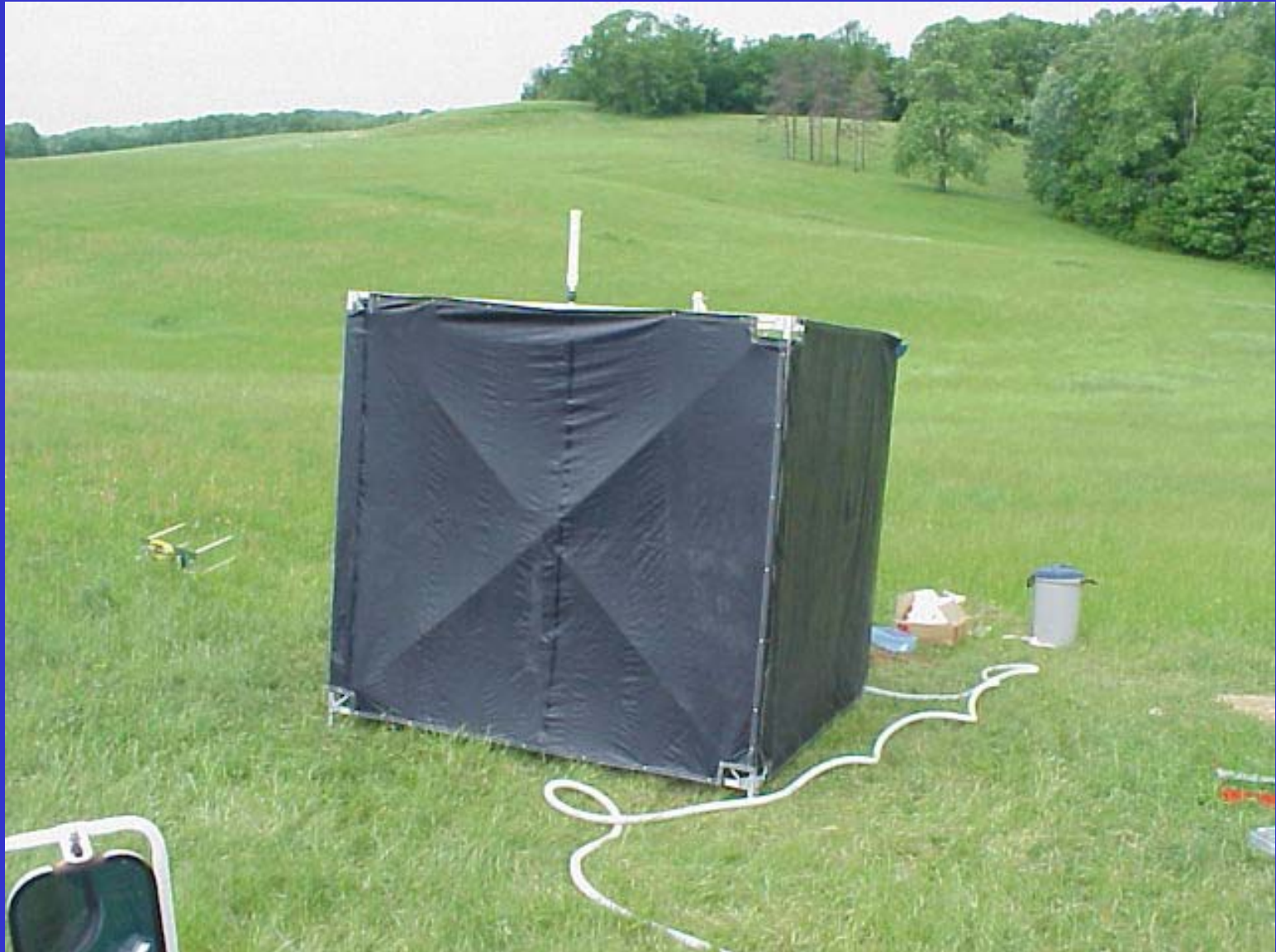
Effectiveness Of Border Strips Treated With AMD Floc In Controlling Runoff P Loss

- ❖ Rainfall simulation experiment:
 - runoff plots set up at three different locations in Grant and Pendleton counties (WV).
 - 0, 20, 40, and 60 per cent of the area covered by slurry of AMD floc.
 - USDA-ARS rainfall simulator.
 - Four rainfall events

Runoff Plots



Rainfall Simulator



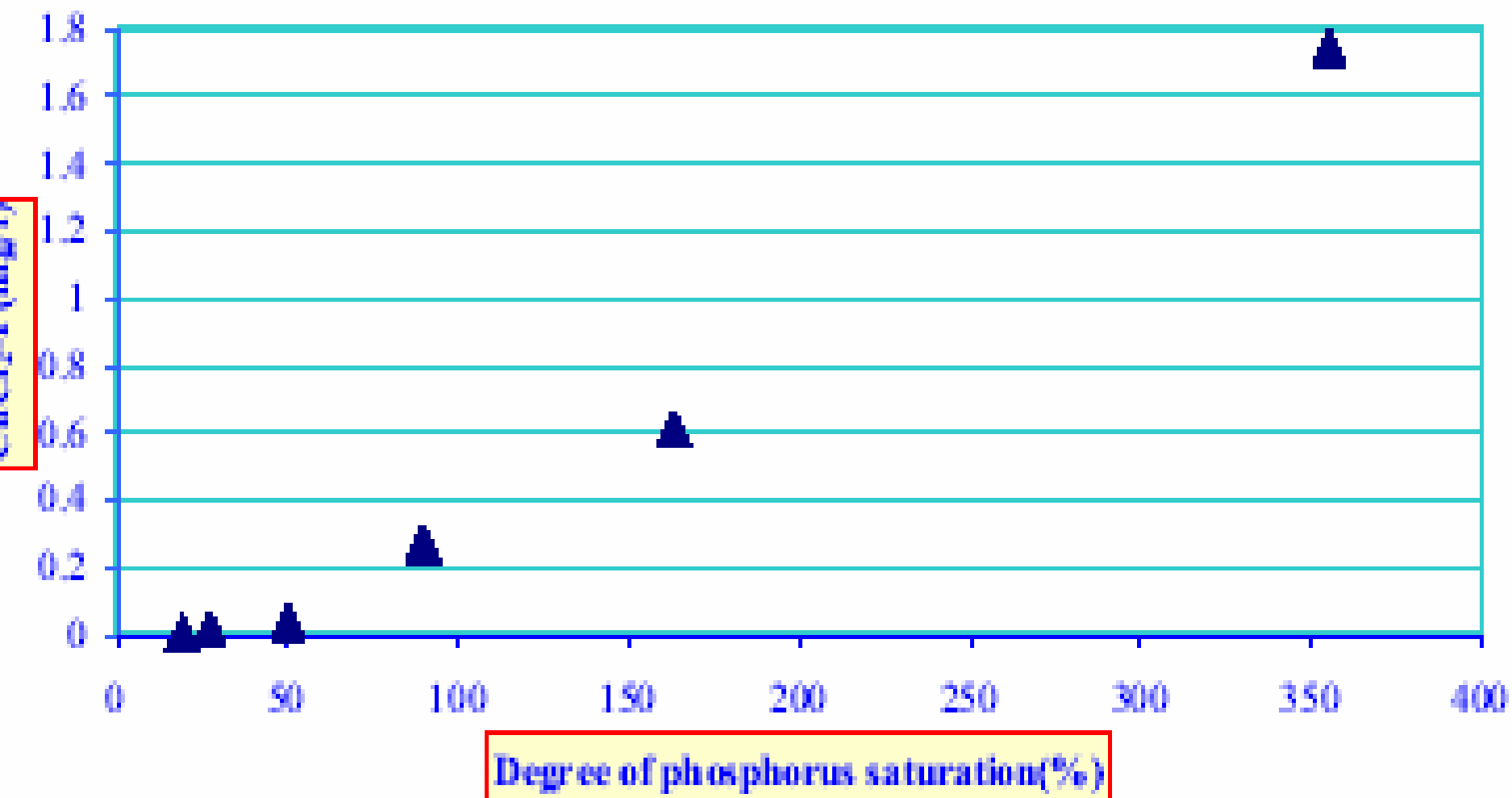


Fig. 8. Effect of acid mine drainage floe additions on Mehlich-1 P.

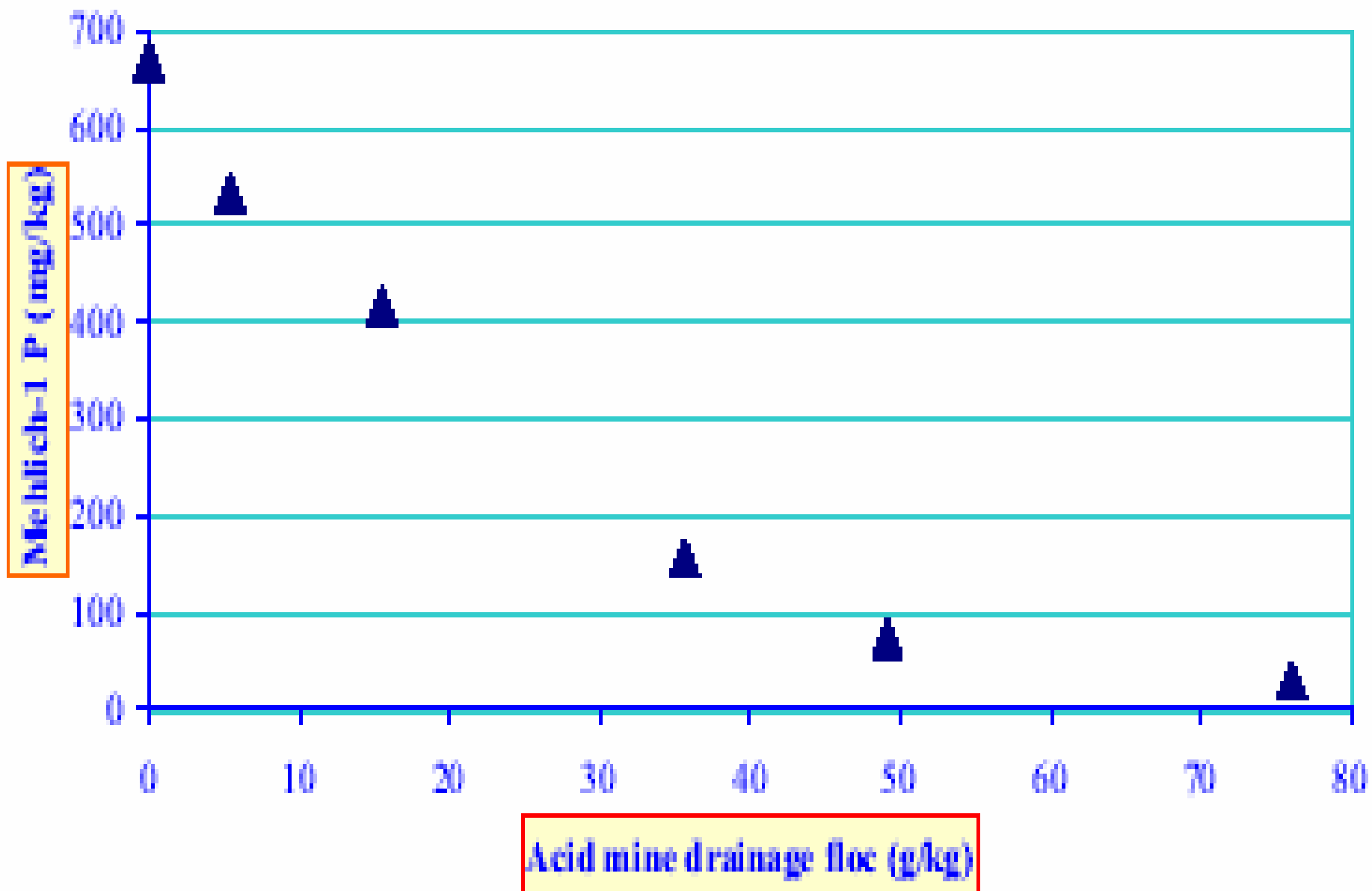
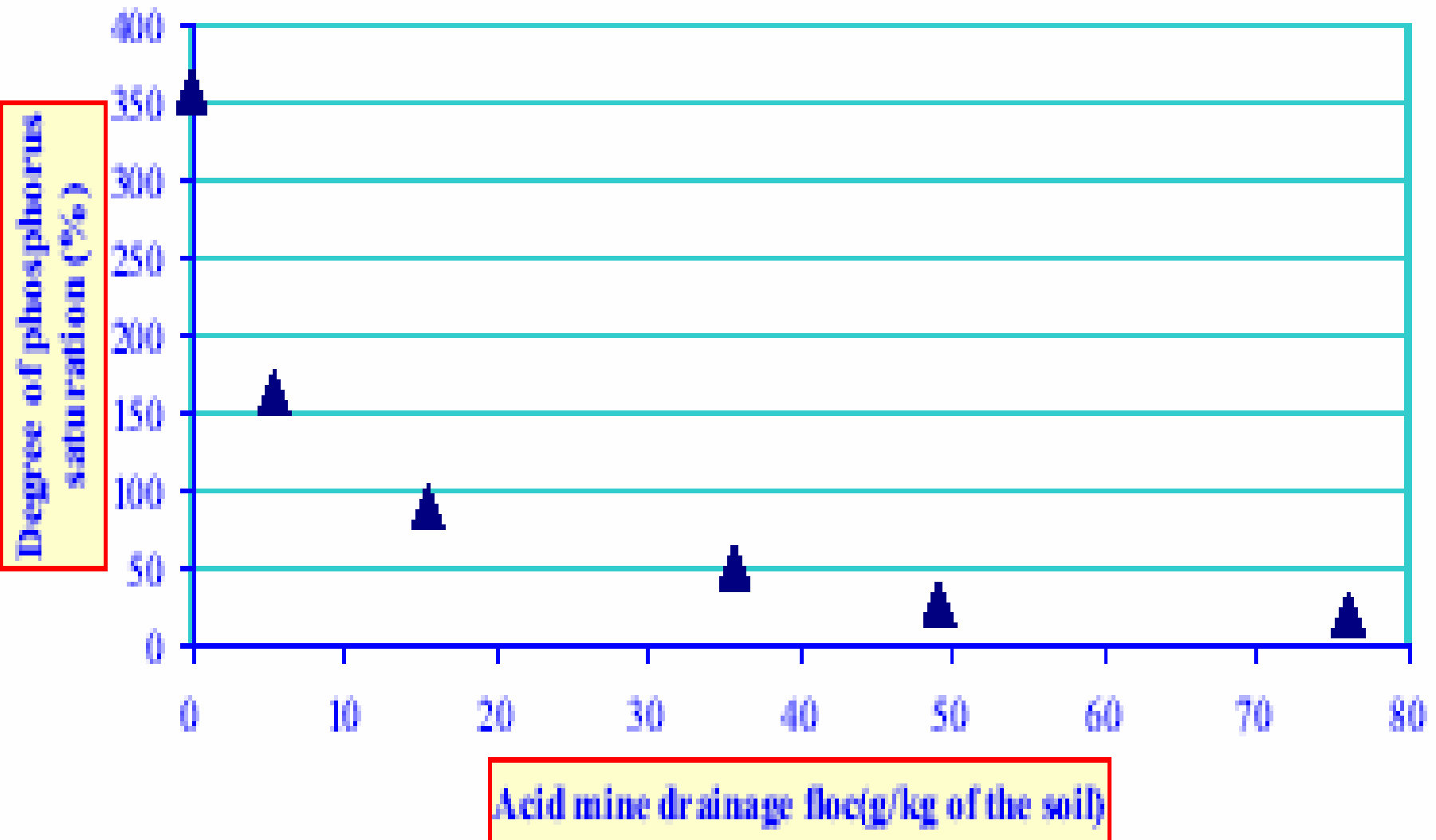


Fig. 10. Effect of acid mine drainage floe additions on degree of phosphorus saturation.



Border Strips

PHOSPHORUS

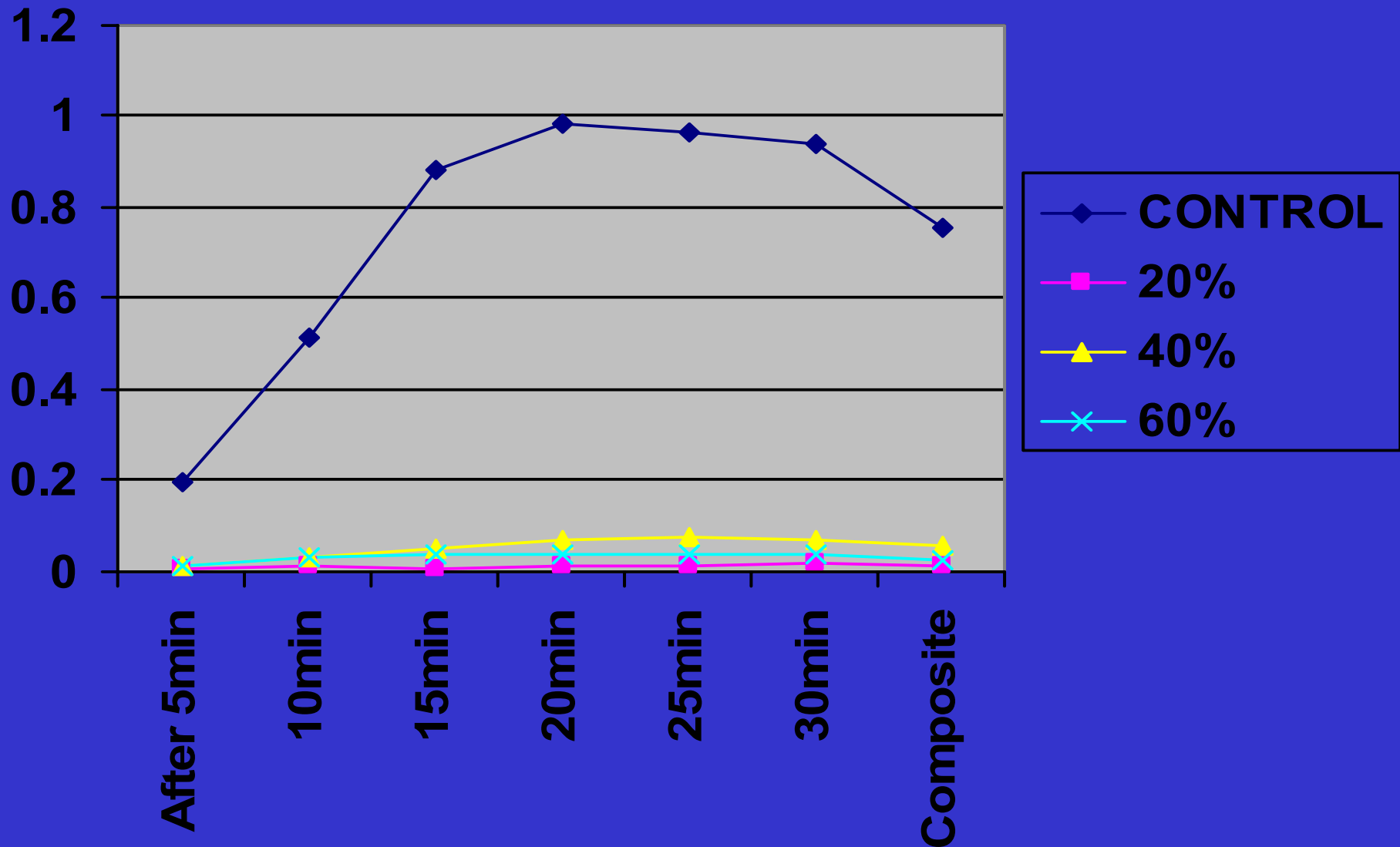


Fig. 13. Effect of acid mine drainage floe buffer strips on dissolved P in runoff from site 1.

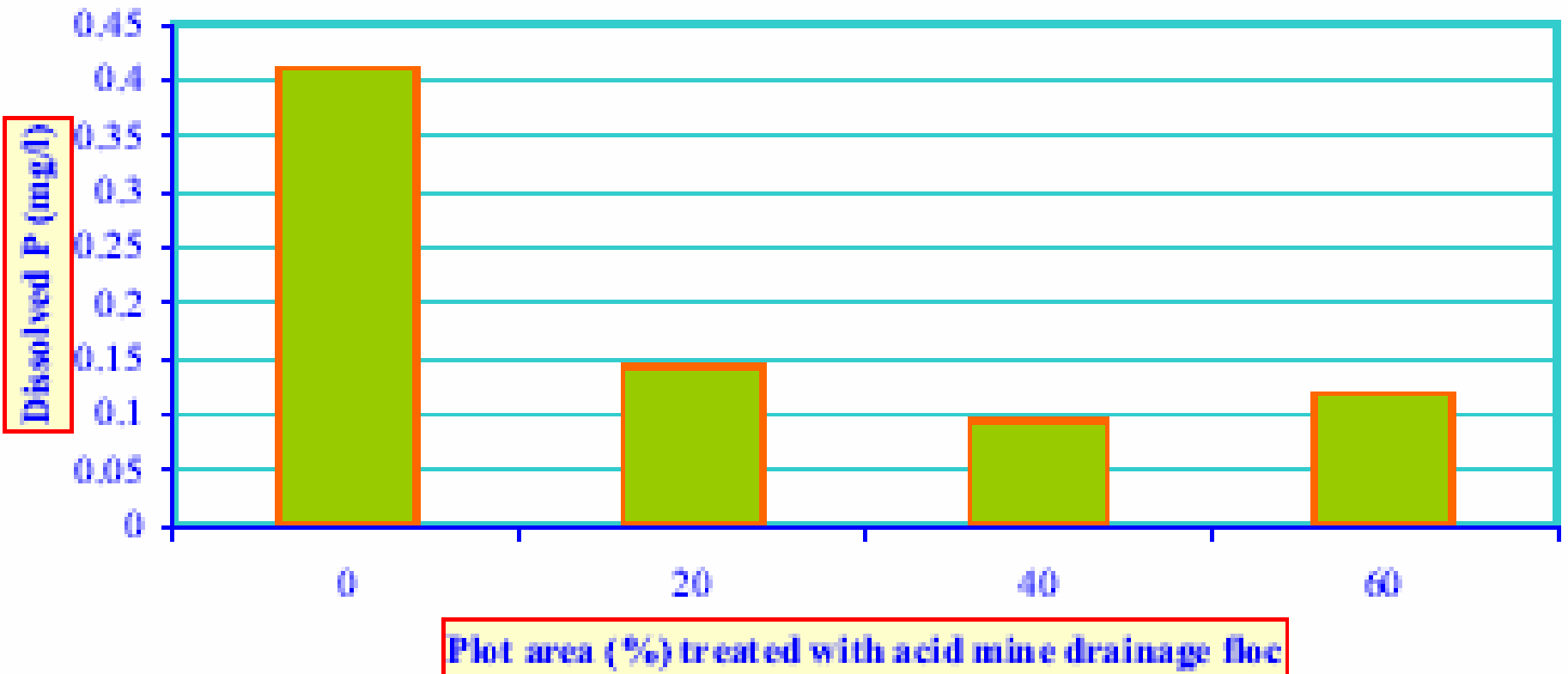


Fig. 14. Effect of acid mine drainage flocc buffer strips on dissolved P in runoff at Site 2.

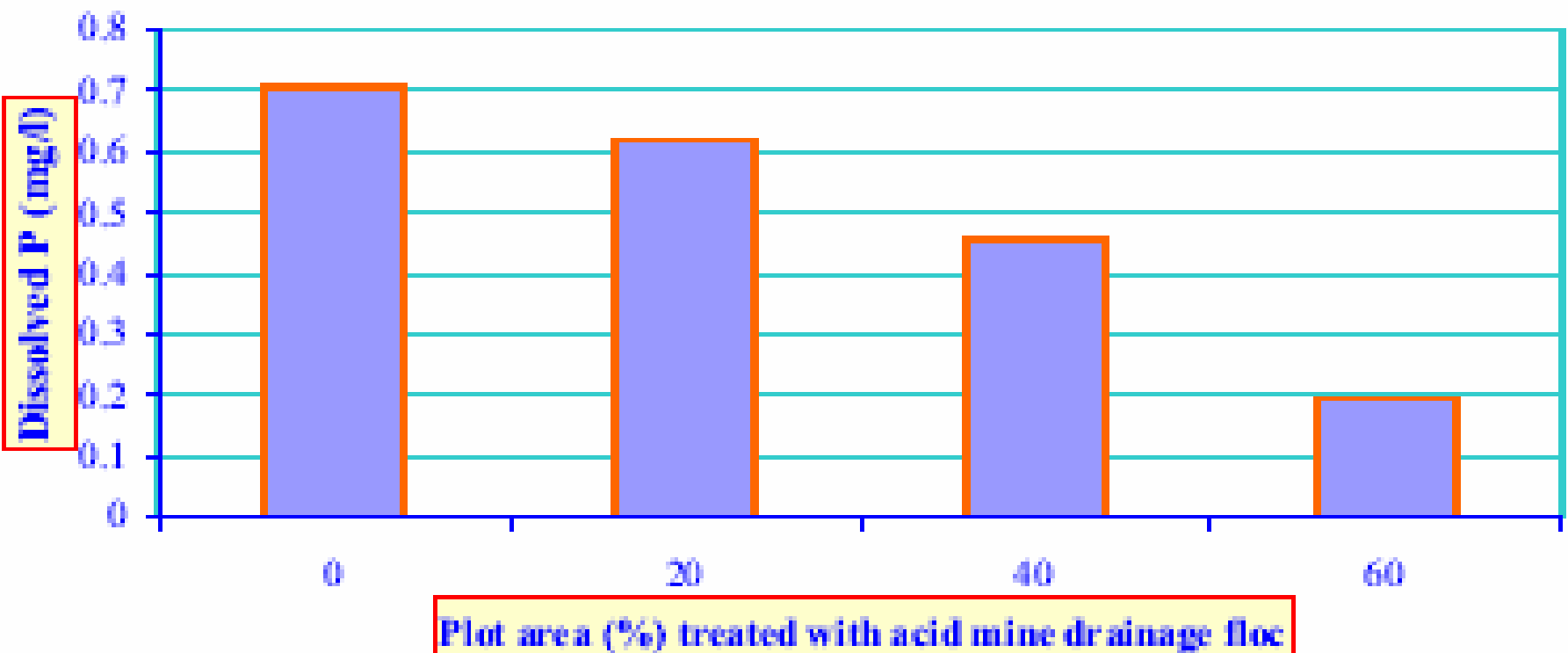
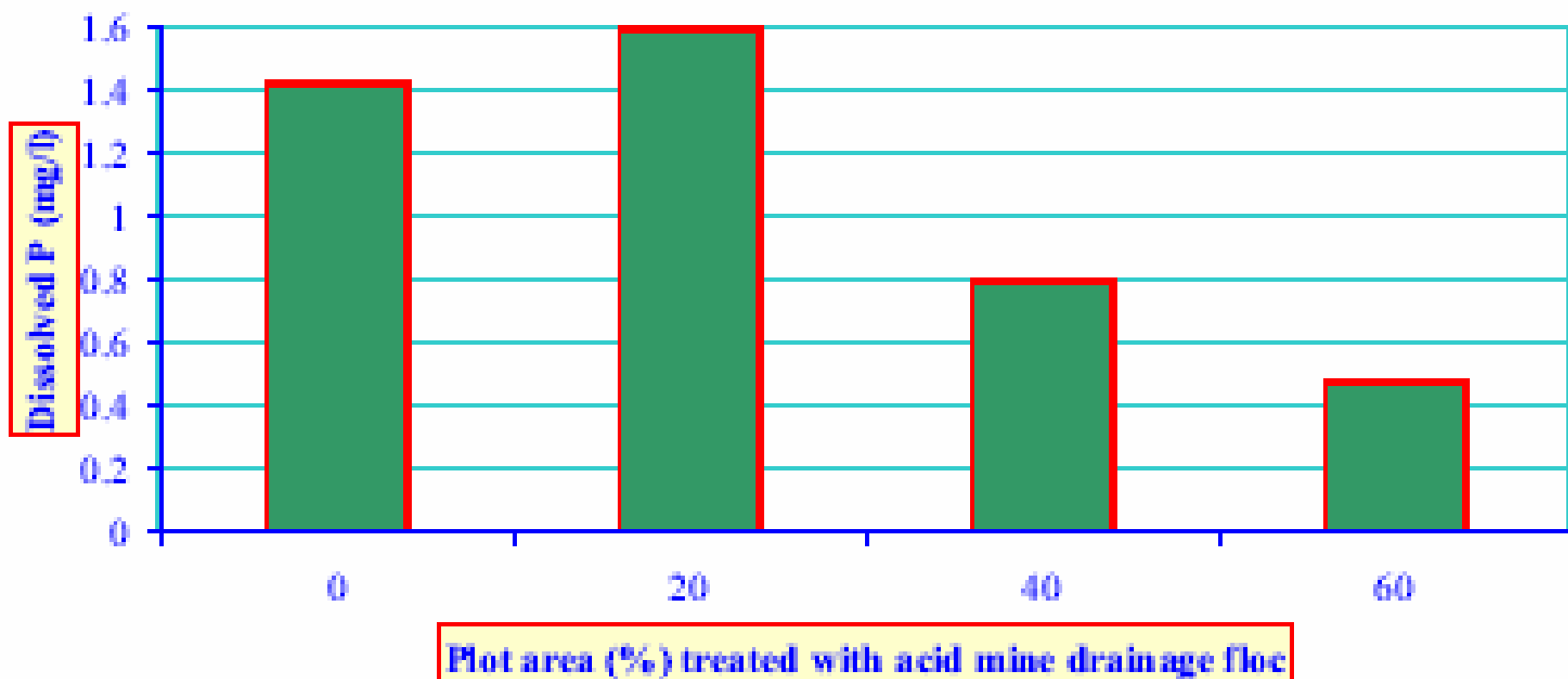
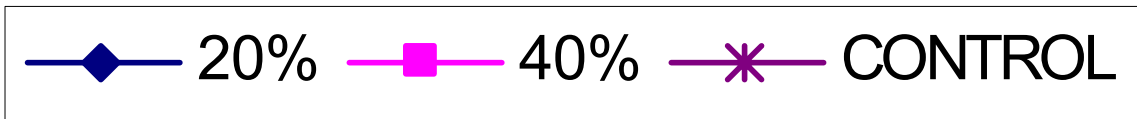
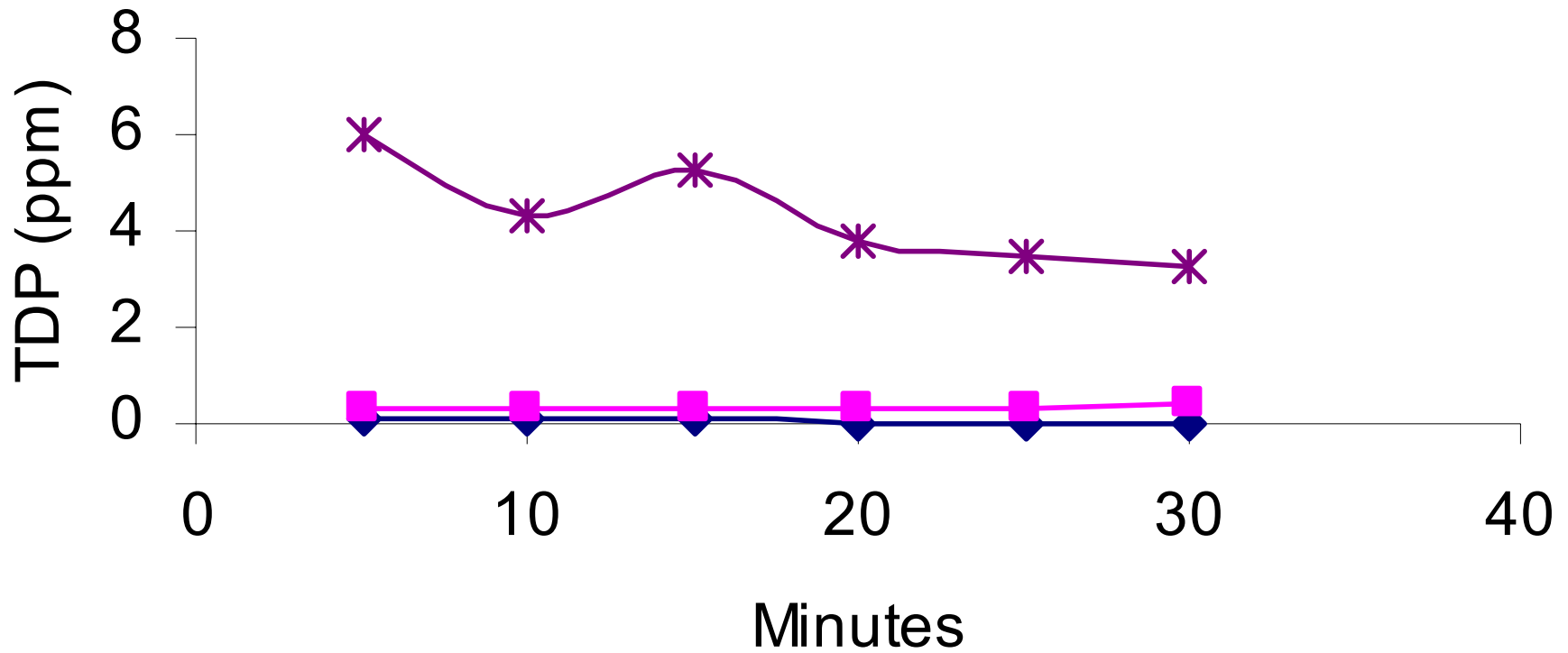


Fig. 15. Effect of acid mine drainage flocc buffer strips on dissolved P in runoff at Site 3.

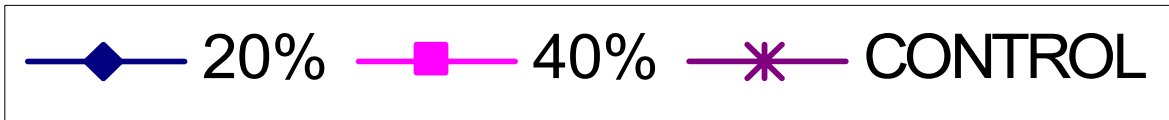
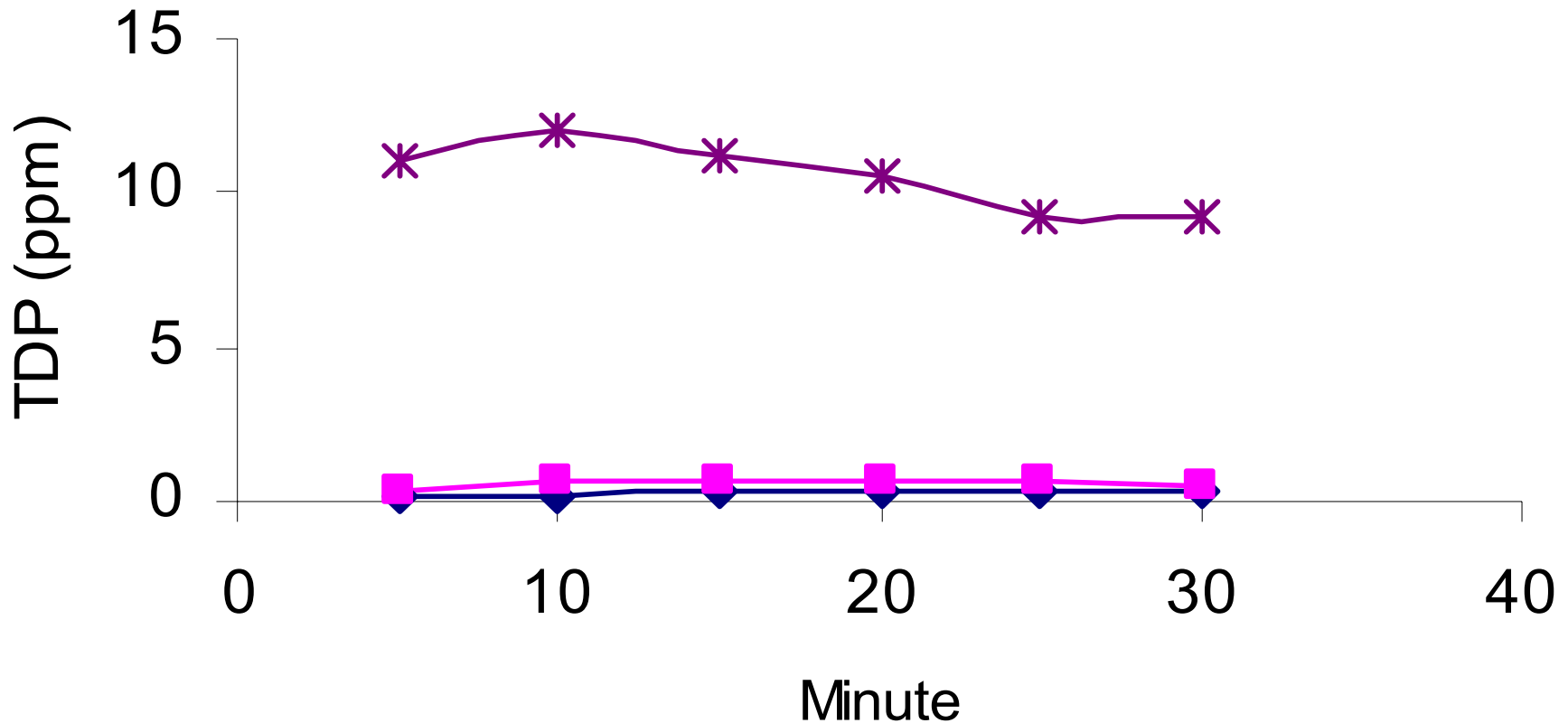


Manure Storage Areas

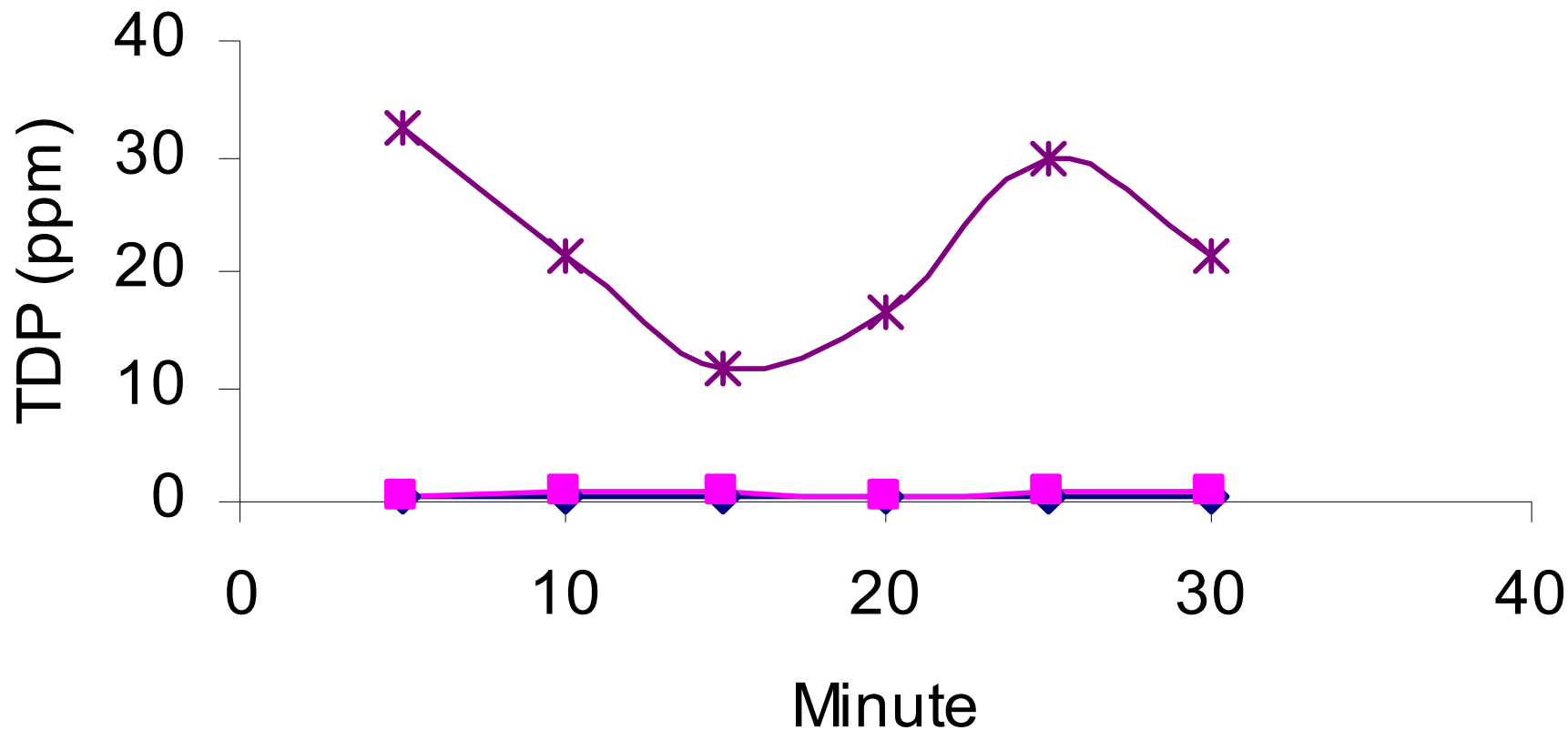
Total Dissolved Phosphorus In Runoff At 2% Slope



Total Dissolved Phosphorus In Runoff At 8% Slope



Total Dissolved Phosphorus In Runoff At 16% Slope



—◆— 20% —■— 40% —*— CONTROL

Conclusions

Acid mine drainage floc has very high capacity to fix P

High amounts of phytic acid phosphate is fixed by AMD flocs

Border strips treated with AMD floc reduce dissolved P

AMD floc reduces dissolved P from temporary manure storage areas

Acknowledgements

- WVSCA
- Rosa Lee Miller, Matt Stroud, and Tom Basden